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(54) 3-AMINOISOTHIAZOLO-[3,4-d]-PYRIMIDINES

We, BASF AKTIENGESELL-SCHAFT, a German Joint Stock Company of 6700 Ludwigshafen, Federal Republic of Germany, do hereby declare the invention, for which we pray that a Patent may be granted to us, and the method by which it is to be performed to be particularly described in and by the following statement:-

The present invention relates to 3-aminoisothiazolo-[3,4-d]-pyrimidines and to a process for preparing such 3-aminoisothiazolo-

[3,4-d]-pyrimidines.

The ring system of isothiazolo-[3,4-d]-pyrimidines was described for the first time by K. Hartke and L. Peshkar in Archiv der Pharmazie, 301, 611 to 621. Starting from malononitrile and dithio or thio esters there are obtained in a condensation reaction the salts of 2-mercapto-1-cyano-acrylonitriles which by reaction with chloramine give 3amino - 4 - isothiazolecarbonitriles; the latter there are obtained by reaction with orthoformic acid esters and acetic anhydride 3 - [ethoxymethylene] - amino - 4 - isothiazolecarbonitriles which can be cyclized with ammonia to give isothiazolo-[3,4-d]pyrimidines.

We have now found 3-aminoisothiazolo-[3,4-d] - pyrimidines formula: of the general

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in which R1 and R2 are, each independently, hydrogen, a linear or branched, saturated or unsaturated alkyl group of one to six carbon 35 atoms which optionally may be substituted, an aralkyl or unsubstituted or substituted cycloaliphatic, heterocyclic or aromatic radical, and Ra is hydrogen, sulfonyl or carboxylic acyl.

Examples of alkyl for R1 and R2 are

methyl, ethyl, n-propyl, isopropyl, n-butyl, isobutyl, and n-hexyl; examples of unsaturated alkyl are allyl or methallyl and examples of substituted alkyl are γ -methoxypropyl, tert-butoxypropyl and β -dimethylaminoethyl.

Benzyl, phenylethyl and γ-phenylpropyl are suitable aralkyls.

The cycloaliphatic radicals are preferably derived from cycloaliphatic rings having three to six carbon atoms in the ring and may be substituted by alkyl such as methyl or ethyl. Cyclohexyl or cyclohexyl bearing methyl as a substituent are particularly suitable.

Examples of heterocyclic radicals are the furyl and thienyl radicals which may be sub-

stituted in conventional manner.

The preferred aromatic radical is the phenyl group which may bear chloro, bromo, methoxy, ethoxy, methyl or ethyl as substituents, such as p-chlorophenyl, p-methylphenyl and

p-methoxyphenyl.

Suitable sulfonyl radicals for R³ include Suitable sulfonyl radicals for R³ include alkylsulfonyl, preferably of one to four carbon atoms, cycloalkylsulfonyl, unsubstituted or substituted arylsulfonyl, aralkylsulfonyl or heteroarylsulfonyl. Specific examples are methylsulfonyl, ethylsulfonyl, propylsulfonyl, butylsulfonyl, cyclohexylsulfonyl, benzylsulfonyl, phenylsulfonyl, chlorophenylsulfonyl, dichlorophenylsulfonyl, nitrorhenylsulfonyl, methylphenylsulfonyl, dichlorophenylsulfonyl, dichlorophen nitrophenylsulfonyl, methylphenylsulfonyl, di-

methylphenylsulfonyl and thienylsulfonyl.

Carboxylic acyl radicals for R^s include substituted or unsubstituted alkanoyl radicals preferably of one to seven carbon atoms, such as acetyl, propionyl, butyryl, caproyl, hexanoyl, capryloyl, \(\beta\)-ethylhexanoyl, chloroacetyl, bromoacetyl, \(\alpha\)-chloropropionyl, \(\beta\)chloropropionyl, γ -chloro-n-butyryl and γ -bromoisovaleroyl; aralkanoyl radicals such as phenylacetyl, tolylacetyl, methoxyphenylacetal, chlorophenylacetyl or dichlorophenylacetal, alkoxycarbonyl or aroxycarbonyl radicals such as methoxycarbonyl, ethoxycarbonyl, propoxycarbonyl, butoxycarbonyl, or phenoxycarbonyl, substituted or unsubstituted aroyl, such as

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benzoyl, or benzoyl, naphthoyl or phenanthronyl bearing chloro, bromo, nitro, methoxy, ethoxy or methyl as substituents, or a heterocyclic radical such as furoyl or thienoyl.

The invention also relates to a process for the production of 3-aminoisothiazolo-[3,4-d]-pyrimidines of formula I wherein a 4-amino-uracil-5-carboxylic thioamide of the formula (II):

in which R^1 , R^2 and R^3 have the above meanings is

oxidatively cyclized.

4 - Aminouracil - 5 - carboxylic thioamides of the formula (II) may be prepared
in accordance with German Patent Specification 21 50 686 from 4-aminouracils by reaction with a sulfonyl or carboxylic acyl isothiocyanate according to the equation:—.

$$0 \xrightarrow{R^{1}-N} N^{1} \xrightarrow{R^{2}-R=C=S} \longrightarrow (II)$$

where R¹, R² and R³ are the corresponding substituents. Reaction of the 4-aminouracils (which are prepared by prior art methods, for example J. Org. Chem. 16 (1951), 1879 to 1890) with a sulfonyl or carboxylic acyl isothiocyanate which are also prepared by prior art methods for example German Printed Application DAS 1,183,492 and Houben-Weyl, "Methoden der organischen Chemie", 4th edition, 9, 879, G Thieme, Stuttgart, 1955, and J. Org. Chemistry, 29 (1964), 2261 and Phannaz. Zentralhalle 197, (1968). page 277) is carried out at temperatures of from 0° to 200° C expediently in a solvent and the reaction product is processed by a conventional method.

The reaction of the 4-aminouracil with the

The reaction of the 4-aminouracil with the sulfonyl or carboxylic acyl isothiocyanate is 0 carried out conveniently at temperatures of from 10° to 200° C, preferably from 20° to 100° C, in a solvent or diluent inert to the reactants. These include for example 1,2-dimethoxyethane, tetrahydrofuran, dioxane, formamide, dimethylformamide, dimethylacetamide, N-methylpyrrolidine, tetramethylenesulfone, benzene, toluene, chlorobenzene, benzonitrile and acetonitrile. The reaction may also be carried out without an added solvent using an excess of allyl isothiocyanate.

For example 0.1 mole of a 4 aminouracil of the said formula may be dissolved or suspended at ambient temperature in a solvent

or diluent and then while stirring at least 0.1 mole of sulfonyl or carboxylic acyl isothiocyanate is added. Almost invariably an exothermic reaction commences immediately. To complete the reaction the whole is heated for from half an hour to three hours, preferably from one hour to two hours, at 40° to 80° C. The reaction product, which often is sparingly soluble, usually crystallizes out on cooling; if necessary some of the solvent may be distilled off at subatmospheric pressure or ether, ligroin, cyclohexane or water may be added to the concentrated solution in order to achieve or complete the separation.

The reaction mixture is processed by a conventional method, for example by suction filtration of the precipitate, washing for example with water, an alcohol, an ether, ligroin or cyclohexane, and drying at subatmospheric pressure.

When carrying out the reaction with an excess of allyl isothiocyanate (the 4-amino-uracil being introduced for example into about 4 to 8 times the molar amount of the same advantageously with stirring and after the exothermic reaction has subsided the whole is heated for another one hour to three hours at 40° to 80° C) it is advantageous to use for the isolation of the reaction product a diluent which does not dissolve it, for example an aliphatic or cycloaliphatic hydrocarbon. The reaction product is then obtained in a form in which it can easily be suction filtered; unreacted isothiocyanate can be recovered from the filtrate by fractionation.

Oxidative cyclization of the 4-aminouracil-5-carboxylic thioamides is conveniently carried out in the presence of an inert solvent and/or suspension agent at a temperature of from 0° to 180° C preferably at from 10° to 120° C.

Examples of oxidizing agents which effect cyclization are hydrogen peroxide and derivatives thereof, e.g. Caro's acid, persulfuric acid, peracetic acid, perboric acid, perphthalic acid or perbenzoic acid, bromine, chlorine, sulfuryl chloride, sulfur trioxide, sulfur ditloride, chromic acid, hypochlorous acid, perchloric acid, ozone or concentrated sulfuric acid.

Solvents and/or suspension agents which are inert to the reactants are conveniently 105 used. Thus for example water, lower fatty acids, alkanols, ethers, halohydrocarbons, lower fatty acid esters, amides of lower fatty acids and mineral acids may be used as solvents. The process may also be carried 110 out in the presence of a base, for example pyridine, sodium hydroxide or potassium hydroxide. Suitable reaction media include: acetic acid, propionic acid, chloroform, dimethylformamide, dimethylsulfoxide, water 115 and sulfuric acid.

The preferred oxidizing agents are concentrated sulfuric acid (which in excess may

		3,697	3
	be used as a solvent as well), bromine (par- ticularly in solution in chloroform or di-	o-methylbenzoylisothiocyanate, m-methylbenzoylisothiocyanate,	6.
	methylrormamide) and about 50% hydrogen	p-methylbenzovlisothiocvanate.	
5	peroxide solution using dimethylformamide as an additional solvent.	3,5-dimethylbenzoylisothiocvanate.	
J	To carry out the process the 4-aminouracil-	2,6-dichlorobenzoylisothiocyanate,	
	5-carboxylic acid thioamide is conveniently	2,4,6-trimethylbenzoylisothiocyanate,	70
-	dissolved or suspended in the solvent used	3,4,5-trimethoxybenzoylisothiocyanate, thienoylisothiocyanate,	
	and then the oxidizing agent is added to it.	a-naphthoylisothiocyanate,	
10	at is advantageous to cool the reaction mixture	β -naphthoylisothiocyanate.	
	and if necessary to heat it for a short time after the initial exothermic reaction has sub-	3-phenanthroylisothiocyanate.	75
	sided. For each mole of 4-aminouracil-5-	phenylacetylisothiocyanate,	
	carboxylic thioamide the amount of oxidizing	m-tolylacetylisothiocyanate, p-tolylacetylisothiocyanate,	
15	agent used is generally from one to twenty	m-methoxyphenylacetylisothiocyanate,	
	times and preferably from once to twice the equivalent amount.	p-methoxyphenylacetylisothiocvanate	80
	Processing causes no difficulties and de-	p-chlorophenylacetylisothiocyanate.	•
	pends on the oxidizing agent, solvent and for-	carbomethoxyisothiocyanate,	
20 -	suspension agent used. When an acid is used	carbethoxyisothiocyanate, carbobutoxyisothiocyanate,	
	as the oxidizing agent or reaction medium	phenoxycarbonylisothiocyanate,	85
	of which an oxidizing agent is used which	acetylisothiocyanate.	0.
	forms an acid under the reaction conditions, a convenient procedure is to add the reaction	propionylisothiocyanate.	
25	mature to ice-water and then to add sikeline	butyrylisothiocyanate,	
	SUDSTRUCES Such as caustic sode solution	n-caproylisothiocyanate, n-capryloylisothiocyanate,	~
	Causuc potash solution, aqueous ammonia	chloroacetylisothiocyanate,	90
٠	sodium carbonate or potash until the reaction mixture is neutral. The isothiazolo-[3,4-d]-	a-chloropropionylisothiocyanate.	
30	pyrimidine thus precipitated may be separated	8-chloropropionylisothiocyanate.	
	for instance by filtration or centrifuging	a-bromoisobatyrylisothiocyanate,	00
	when a mineral acid, particularly enforce	γ-chloro-n-butyrylisothiocyanate and α-bromoisovaleroylisothiocyanate.	95
	acid, is used as oxidizing agent and/or ene-		
35	pension agent elimination of the R ³ radical may take place at the same time as oxidative	The 3 - aminoisothiazolo - [3,4-d]-	
	cyclization, a 3-aminoisothiazolo-[3,4-4].	pyrimidines obtained according to the inven	
	pyrimaine mus being obtained which has no	dut are variable intermediates for the pro-	
	substituent on the amine group	duction of dyes, pesticides and medicaments. Compounds in which R ² is hydrogen are	100
40	Naturally the R ² radical may also be eliminated in a separate reaction after cycliza-	particularly important in this respect Hor	
	don has taken blace. Elimination can be of	chample, they are also of interest as notential	
	recied not only with acids, for example out	purine anumetabolites.	
	Turic acid and hydrochloric acid, but also with	The 3 - aminoisothiozolo - [3,4-d]- pyrimidines of the invention have valuable	105
45	bases, for example caustic soda solution and caustic potash solution.	uncrapeutic properties for example of Ai-	
••	Examples of suitable sulfonyl and carboxylic	metics of as anti-inflammation agents and	
	acyl isothiocyanates as starting compounds	mey may also be used as fungicides. Accord.	
	are:—	ingly, the invention includes within its scope pharmaceutical compositions containing a	110
	p-toluenesulfonylisothiocyanate,	movel compound according to the invention	
50	Denzenesulfonvlisothiocomate	wedler with a pharmaceutically-acceptable	
	o-toruenesulforivisothiocvanate	carrier or dillient and also funcicidal ones	
	o-chlorobenzenesulfonvlisothiocuanate	positions comprising such a compound to- gether with a carrier or diluent.	115
	m-chlorobenzenesulfonvlisothiocvanete	The following compounds in particular	
55	p-chlorobenzenesulfonylisothiocyanate, 3,4-dichlorobenzenesulfonylisothiocyanate,	maye a good biological action. 3-amino.5	
	Unophene-2-sulfonvisothiocvanate	/ = Clincinvilsofhiazolo = [3.4.4]	
	Cyclonexanesulfonvlisothiocyanate	uniculone = (4.6). 3 = (N = control or or	120
	methanesulfonvlisothiocvanate.		
60	ethanesulfonylisothiocyanate.	pyrimidinedione - (4,6), 3 - [N - p - tolu- enesulfonyl] - amino - 7 - ethylisothiazolo-	
	benzoylisothiocyanate, 4-methoxybenzoylisothiocyanate,	1257-41 - DVIIIIIIIIIIIIIIIII (4.6) and 2	
	p-chlorobenzoviisothiocyanate	allillo = 2./ - dipropolisothiazolo[-2.4.4]	125
	p-nitrobenzovlisothiocvanate.	Pyrimumiculone-(4.6).	
	m-methoxybenzoylisothiocyanate,	Formulations with the compounds of the	
		invention as active ingredients may be ob-	

tained by conventional methods depending on the type of application required. Particular importance attaches to compounds of the formula (I) in which:

R1 is hydrogen, alkyl of one to four carbon atoms, benzyl, phenylethyl, cyclohexyl or phenyl which may bear chloro, methyl or methoxy as a substituent;

R² denotes, except for hydrogen, the same radicals as R¹ and additionally alkoxyalkyl of a total of up to six carbon atoms; and R² has the above meanings.

For use as diazo components compounds having alkyl as R¹ and R² and particularly methyl, are preferred; R³ is naturally hydrogen in this case.

The following examples, in which parts and percentages are by weight unless otherwise stated, illustrate the invention.

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Example 1.

354 parts of 3 - methyl - 4 - aminouracil-5 - [N - (p - toluene sulfonyl)] - carboxylic thioamide is gradually introduced into 1000 25 parts of concentrated sulfuric acid while stirring at a temperature of about 75° C and in the course of thirty minutes. Reaction immediately takes place as evidenced by the escape of sulfur dioxide. When all has been added the mixture is stirred for another two hours at 80° C. The clear solution is cooled and poured onto about 5000 parts of icewater. The deposited colorless solid is suction filtered, washed with water until neutral and then dried in vacuo at 80° C. Chromatographic and infra-red spectroscopic analysis show the product to be identical with a sample recrystallized from dimethylformamide. The yield is 173.5 parts, which is 88% of theory. Melting point: 360° to 362° C. Analysis: C_eH_eN₄O₂S: molecular weight 198.14. calculated:

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C 36.4 H 3.1 N 28.3 O 16.1 S 16.1 found:

36.3 3.3 27.9 16.3 16.4
The compounds set out as Examples 2 to 7 in Table 1 may be prepared by the process described in Example 1, the sulfonyl or acyl radical being eliminated in each case under the reaction conditions. The following abbreviations are used in Table 1:

Ex=Example No.
Y=Yield
m.p.=melting point
Rec.=recrystallized from
Emp.=empirical formula and molecular
weight
c=calculated
f=found

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5			1,445,697	W		5
		% S	16.1	12.3	15.1	
		% 0	16.1	12.3	15.1	
	Analysis	% N	26.4	21.5	26.4	
		% Н	89 89 89 89	3.1	3.8	
H. S. W. S.		f: C%	39.6	50.8	39.6 39.8	
1 1 0 miles		Emp	C,H,N,O,S 212.17	C ₁₁ H ₈ N ₄ O ₂ S 260.21	dimethyl- C,H,N,O,S formamid 212.17	
		Rec.	dimethyl- formamid	methyl- glycol	dime thyl- formamid	
TABLE 1	•	n.pt.	>330	>300	300–306	
		¥ %	94	66	\$6	
- NH-RS		R,	S E.	805 Sel-Co-	2025 	
		R ₂	сън ,	C,H,	сн,	-
		R,	н	Ħ	CH,	·
		Ex	7	m	4	

		1				
			S) F)s	11.9	10.8	
		20	% 0	11.9	10.8	
		Analysis	2 Z	20.9	18.9	
			ж н	6.0	6.9	
	E To		% ن ن ن	49.3	52.7	
	N N N N N N N N N N N N N N N N N N N		Emp	C ₁₁ H ₁₆ N ₄ O ₂ S 268.27	C ₁₃ H ₂₀ N ₄ O ₂ S 296.32	cf. Example 4
nued)	· •		Rec.	ethanol	ethanol	dimethyl- formamid
TABLE 1 (Continued)			m.pt. °C	208-210	166–169	300308
TABI			≻ 69	97	100	84.9
	R-H-RS		R,	S. E.	% <u></u>	COOC2H5
			Z,	C,H,(n) C,H,(n)	C,H,(n) C,H,(n)	ŧ.
			Ŗ	C,H,(n)	C,H _b (n)	, E
	*		EX	۸.	٥	7

10 15 aqueous ammonia solution, the solid material is suction filtered, washed with water and dried in vacuo at 70° C. The yield is 2.1 parts which is quantitative.

The product is identical with the compound described in Example 4.

The acyl radicals in Examples 9 to 11 (set out in Table 2) may also be eliminated by the process of Example 8.

TABLE 2

Emp. and analysis	as in Example 4	as in Example 4	as in Example 6
Rec.			ethanol
n.pt.	-		167–169
7%	100	95	100
R ₃	8-{\}	соос [;] н ,	85
R ₂	сн,	сн,	C,H,(n)
я.	сн,	CH,	C,H,(n)
Ex .	9	10	=

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Example 12.

6.4 parts of bromine dissolved in 30 parts of chloroform is allowed to drip, with stirring and at 20° to 35° C, into a suspension of 7.1 parts of 3-methyl-4-aminouracil-5-[N-(p-toluenesulfonyl)]-carboxylic thioamide in 100 parts of chloroform. The whole is stirred for another two hours at ambient temperature and the solid is suction filtered. The product is stirred into aqueous ammonia solution, suction filtered, washed with water and dried in vacuo at 70° C. It may be recrystallized from dimethylformamide. The yield is 6.8 parts which is 97% of theory. Melting point: 300° to 305° C. Analysis: C₁₂H₁₂N₄O₄S₂: molecular weight: 352.26.

C 44.3 H 3.4 N 15.9 O 18.2 S 18.2

20 found:

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43.9 3.5 15.4 18.3 17.9.

Example 13.

8.6 parts of 3-benzyl-4-aminouracil-5-[N-(p-tohuenesulfonyl)] - carboxylic thioamide is cyclized as described in Example 12. The solid obtained from the chloroform solution is stirred first in a small amount of ethanol and then with aqueous ammonia solution, suction filtered, washed with water and dried. The yield is 8.5 parts which is quantitative. The melting point is 259° to 264° C. The product may be recrystallized from dimethylformamide. Analysis: C₁₉H₁₆N₄O₄S₂: molecular weight: 428.35. Calculated:

C 53.1 H 3.7 N 13.2 O 15.0 S 15.0 found:

52.9 3.8 13.4 14.6 14.7.

Example 14.

10.4 parts of 1,3-dibenzyl-4-aminouracil-5 - [N - (p - toluenesulfonyl)] - carboxylic thioamide is cyclized as described in Example 12. The product is precipitated from the chloroform solution by adding 100 parts of

ether. It is then suction filtered, washed and dried. The yield is 10.3 parts which is quantitative. It may be recrystallized from methyl glycol. The melting point is 230° to 231° C. Analysis C₂₆H₂₂N₄O₄S₂: molecular weight 518.47. Calculated:

C 60.2 H 4.3 N 10.8 O 12.3 S 12.3 found:

60.1 4.4 10.9 12.4 12.2.

Example 15.

9.6 parts of bromine is dripped, while stirring and at 20° to 40° C, into 13.6 parts of 1,3 - di - n - butyl - 4 - aminouracil - 5- [N - (p - toluenesulfonyl)] - carboxylic thioamide in 100 parts of chloroform. The solution is stirred for another three hours at ambient temperature and then 200 parts of ether is added. The solid is suction filtered and dried in vacuo. The yield is 11.8 parts, which is 88% of theory. It can be recrystallized from methyl glycol. The melting point is 183° to 185° C. Analysis: $C_{20}H_{28}N_4O_4S_2$: molecular weight: 450.44. Calculated:

C 53.3 H 5.8 N 12.4 O 14.2 S 14.2 found:

53.4 5.6 12.2 14.1 14.0 75

Example 16.

32 parts of bromine is dripped, while stirring and at 20° to 35° C, into 28.6 parts of 1,3 - dimethyl - 4 - aminouracil - 5- (N - carbethoxy) - carboxylic thioamide in 400 parts of chloroform. Stirring is continued for another two hours. Almost all temporarily passes into solution; gradually however a thick precipitate settles our. This is suction filtered, dissolved in about 150 parts of dimethyl formamide and then precipitated again with aqueous ammonia solution. The product is suction filtered, washed with water and dried in vacuo at 70° C. The yield is 21.4 parts or 75% of theory. The product may be recrystallized from ethanol or methyl glycol. Melting point: 143° to 145° C. Analysis: C₁₀H₁₂N₄O₄S; molecular weight: 284.23.

Calculated:

C 42.3 H 4.3 N 19.7 O 22.6 S 11.3 found:

42.7 4.3 19.5 23.1 10.6.

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Example 17.

9.1 parts of bromine is allowed to drip at 20° to 40° C and while stirring into 10 parts of 1,3 - dimethyl - 4 - aminouracil - 510 [N - (p - chlorobenzoyl)] - carboxylic

thioamide in 75 parts of dimethylformamide. The whole is stirred for another three hours at ambient temperature, and diluted with water, some aqueous ammonia solution is added, and the solid product is suction filtered, washed with water and dried in vacuo at 70° C. The yield is 30 parts which is 76.8% of theory. It may be recrystallized from dimethylformamide. The melting point is 246° to 250° C. Analysis: C₁₄H₁₁N₄O₅SCI; molecular weight: 350.7. Calculated:

C 47.9 H 3.2 N 16.0 O 13.7 Cl 10.1 found:

48.1 3.4 15.8 13.7 10.6.

The compounds set out in Examples 18 to 24 in Table 3 may be prepared by the process described in Example 17.

			SS:	10.1	17.5	8.0
		Analysis	%0	15.2	17.5	12.0
		Ā	Z %	17.7	15.3	14.0
			H %	8. 8. 8. 8.	3.8	6.2
	F-5-2		:: :: C %	53.2	45.9	60.0
	4. 400	-	Emp,	C ₁₄ H ₁₂ N ₄ O ₅ S 316.27	C,4H,,N,O,S, 366.28	C ₂₀ H ₂₄ N ₄ O ₃ S 400.42
TABLE 3	· •		Rec.	dimethyl- formamid	216-218 glycol	160-162 methyl- glycol
TAB			o. P. P.	ab 220 Zers.	216–218	160–162
	·	:	>> 8€	92	83.1	84
	2 - 2 - 2 - 2 - 2 - 2 - 2 - 2 - 2 - 2 -		R,	≈ √}	g	\$- ()
	or party		R,	_.	CH,	C ₄ H ₆ (n)
	1	-	S.	£	CH,	C ₄ H _p (n)
			Ex	. 18	19	20

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		\$-6	NH PASS			^	1. J. D.	A A A A A A A A A A A A A A A A A A A	r.			
					-					Analysis		
Ex.	쬬.	ጜ	R.	> %	o. D.	Rec.	Emp.	c: f: C%	% Н	Z %	% 0	% %
21	C,H,(n)	C ₃ H,(n) C ₃ H,(n)	88 <u>, </u>	76	171–173	171-173 methyl- glycol	C, H ₂ N, O, S ₂ 422.39	51.2	5.3 5.6	13.3	15.1	
22	. H	(СН,),0СН,	0%-(100	268–270	dimethyl- formamid	C,eH,eN,O,S,	46.8	4.4	13.7	19.5	15.6
. 23	ж	(CH ₂), OC ₂ H ₄	302 OH ₃	89.6	89.6 265-268 glycol	methyl- glycol	C,,H _{2,0} N ₄ O _s S ₂ 424.36	48.1	4.4	13.2	18.9	14.1
24	Ж	(сн.), осн(сн.),		86.5	86.5 258-261 methyl-	methyl- glycol	C ₁₆ H ₁₂ N ₄ O ₁ S ₂ 438.39	49.3	5.1	12.8	18.2	14.6

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Example 25.

A mixture of 4.24 parts of 1,3-dimethyl-4 - aminouracil - 5 - [N - (p - toluene-sulfonyl)]-carboxylic thioamide, 40 parts of dimethylformamide and 2 parts of 50% hydrogen peroxide is heated first for two hours at 50° C and then for another two hours at 100° C. The relative is cooled and water is 100° C. The solution is cooled and water is added. The product which is precipitated is suction filtered. Chromatographically and spectroscopically it is identical with the product described in Example 21. The yield is 2.7 parts, which is 62.5% of theory.

WHAT WE CLAIM IS:—

1. A 3 - aminoisothiazolo - [3,4-d]-15 pyrimidine of the general formula:-

in which R1 and R2 are, each independently, hydrogen, a linear or branched, saturated or unsaturated alkyl group of one to six carbon atoms which optionally may be substituted, an aralkyl or an unsubstituted or substituted cycloaliphatic, heterocyclic or aromatic radical, and R3 is hydrogen, sulfonyl or carboxylic acyl.

2. A compound as claimed in claim 1 and as individually identified in the foregoing Examples.

[3,4-d] - pyrimidinedione - (4,6).

4. 3 - [N - Carbethoxy] - amino - 5,7-dimethylisothiazolo - [3,4-d] - pyrimidinedione-(4,6).

5. 3 - [N - p - Toluenesulfonyl] - amino-7 - ethylisothiazolo - [3,4-d] - pyrimidinedion-(4,6).
6. 3 - Amino - 5,7 - dipropylisothiazolo-

[3,4-d]-pyrimidine-dione-(4,6).

7. A process for the production of a 3-aminoisothiazolo - [3,4-d] - pyrimidine as 40 defined in claim 1 wherein a 4-aminouracil-5-carboxylic thioamide of the formula:---

in which R1, R2 and R3 are each as defined in claim 1, is oxidatively cyclized.

8. A process as claimed in claim 7 wherein the oxidative cyclization is carried out in the presence of an inert solvent and/or suspension agent at a temperature of from 0° to 180° C.

9. A compound as defined in claim 1 and whenever prepared by a process as claimed in claim 7 or claim 8.

10. A pharmaceutical composition comprising a compound as claimed in any one of claims 1 to 6 or 9 together with a pharmaceutically-acceptable carrier or diluent.

11. A fungicidal composition comprising a compound as claimed in any one of claims 1 to 6 or 9 together with a carrier or diluent.

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